Use the vocabulary words and definitions below as a reference for this unit.

atom the smallest unit of an element that is still that element; the basic building block of matter **chain reaction**...... a self-sustaining nuclear reaction; it continues without the addition of outside energies **chemical energy** the energy that is stored in chemicals control rod a barrier that slows a nuclear reaction by absorbing excess radiation **electromagnetic energy** the energy that results from the interaction of the electric and magnetic fields **electromagnetic force** the forces of attraction and repulsion between charged particles, resulting in electricity and magnetism **electron** the negatively charged particle of an atom: the electron moves around the center of the atom (nucleus) energy the ability to do work or cause change **fission** splitting the nucleus of an atom into two lighter parts

fission reactor	a type of nuclear reactor that splits the nuclei of atoms
fusion	a nuclear reaction in which two or more nuclei are pushed together to form one large nucleus
fusion reactor	a type of nuclear reactor that would combine atoms
gravity	the attraction of matter toward another body of matter <i>Example</i> : Earth's gravity holds us on its surface.
half-life□	the time it takes one-half of the atoms of a radioactive sample to decay
isotope	an atom or group of atoms with the same atomic number but different atomic mass than other atoms of a specific element; this difference in mass is based on a difference in the number of neutrons within the nucleus of the atom
law of conservation of energy	the law that energy cannot be created or destroyed, only changed from one form to another during a physical or chemical change
law of conservation of mass	the law that matter cannot be created or destroyed, only changed from one form to another during a physical or chemical change

mass the amount of matter in a substance matter anything that has both mass and volume **mechanical energy** the energy of moving things **neutron** the neutral particle found in the nucleus of an atom; a neutron has no charge **nuclear energy** the energy that holds the nuclei of atoms together; it is released in nuclear reactions and may be used to produce heat, electricity, or other forms of energy **nuclear reaction** a reaction that occurs when an atom is split; large amounts of energy are released **nuclear reactor** a machine used to control or create a nuclear chain reaction nucleus the center region of the atom around which the electron(s) move; plural: nuclei **proton** the positively charged particle in the nucleus of an atom radiation the movement of energy as a wave radioactive describing those elements or isotopes that spontaneously decompose and give off radiation

radioactive waste the waste produced by a nuclear reactor; though unusable it still releases radiation

radioactivity forms of energy given off by nuclear material

theory of relativity the theory that there is a fundamental relationship between matter and energy; $E=mc^2$ (E stands for energy, m stands for mass, and c stands for the speed of light.)